

### [tex13] Room heater: Electric radiator versus heat pump

A room is to be kept at temperature  $T_H = 294\text{K}$ , ( $21^\circ\text{C}$ ). The outdoor temperature is  $T_L$ . Heat, which leaks through the windows and doors at the rate  $\bar{Q}_{leak} = \gamma\Delta T$ , must be replaced by a room heater at the same rate. The electric radiator converts electric power  $\bar{W}_{el}$  into heat with 100% efficiency. The electric heat pump uses the amount  $\bar{W}_{sup}$  of electric power to drive a Carnot cycle in the reverse, which extracts heat  $\bar{Q}_L$  at temperature  $T_L$  from the exterior and converts it (reversibly) into heat  $\bar{Q}_H = \bar{Q}_L + \bar{W}_{hp}$  at temperature  $T_H$ . In the relation  $\bar{W}_{hp} = (1 - \lambda)\bar{W}_{sup}$ ,  $\lambda$  represents the energy loss in the gears of the heat pump. Quantities with overbars denote energy transfers per time unit.

- (a) Find  $\bar{W}_{el}$  as a function of  $\gamma, T_H, T_L$ , and  $\bar{W}_{sup}$  as a function of  $\gamma, \lambda, T_H, T_L$ .
- (b) Plot  $\bar{W}_{el}/\gamma$  and  $\bar{W}_{sup}/\gamma$  versus  $t_L \equiv T_L - 273\text{K}$  (measured in  $^\circ\text{C}$ ) for fixed  $T_H = 294\text{K}$  and  $\lambda = 0.8$  (20% efficiency).
- (c) Determine the range of  $T_L$  where the heat pump is more economical than the radiator.

**Solution:**